

## After Dok Phys-271

Finding time between collisions in a cold molecular Hydrogen cloud, with a molecule density of

$n = 8.33 \times 10^8 \text{ mol/m}^3$ , a temperature of  $T = 15 \text{ K}$ .

\* Approx  $d_{\text{H}_2} = 1.6 \times 10^{-10} \text{ m}$

$$\text{Solution: } mfp = \frac{1}{\pi d^2 n} \rightarrow \frac{1}{\pi (1.6 \times 10^{-10} \text{ m})^2 (8.33 \times 10^8 \text{ mol/m}^3)}$$

$$mfp = 1.49 \times 10^9 \text{ m} \rightarrow 0.0998 \text{ AU}$$

$$v_{\text{H}_2} = \sqrt{\frac{3KT}{2m_{\text{H}_2}}} \rightarrow T = 15 \text{ K}, m_{\text{H}_2} = 2m_{\text{H}} = 2 \cdot (1.67 \times 10^{-27} \text{ kg})$$

$$v_{\text{H}_2} = \sqrt{\frac{3(1.38 \times 10^{-23} \text{ J/K})15 \text{ K}}{3.34 \times 10^{-27} \text{ kg}}} = 431.2 \text{ m/s}$$

$$t = \frac{mfp}{v} = \frac{1.49 \times 10^9 \text{ m}}{431.2 \text{ m/s}} = 3.46 \times 10^7 \text{ s}, 399.99 \text{ d}, 1.096 \text{ yr}$$