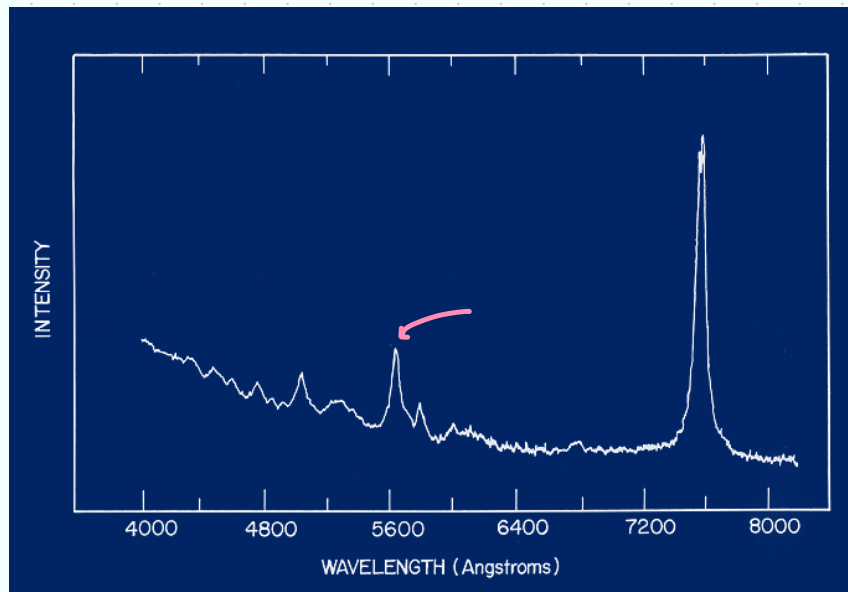


Student Question - Priti Finavia



The arrow shows an emission line of an object. The redshift of this object is 0.152

A) Find rest wavelength of the emission line.

$$\Rightarrow z = \frac{\lambda_{\text{obs}} - \lambda_{\text{rest}}}{\lambda_{\text{rest}}}$$

$$0.152 = \frac{5600 \text{ \AA} - \lambda_{\text{rest}}}{\lambda_{\text{rest}}} \Rightarrow \lambda_{\text{rest}} = 4861.1 \text{ \AA}$$

B) What transition does this wavelength correspond to, assuming it is hydrogen line.

$$\Rightarrow E = \frac{hc}{\lambda} = \frac{1240 \text{ eV nm}}{486.11 \text{ nm}} = 2.55 \text{ eV}$$

$$E = -13.6 \text{ eV} \frac{1}{n^2}$$

$$E_1 = -13.6 \text{ eV} \quad E_3 = -1.51 \text{ eV}$$

$$E_2 = -3.4 \text{ eV} \quad E_4 = -0.85 \text{ eV}$$

$$\text{so, } E_4 - E_2 = -0.85 + 3.4 = 2.55 \text{ eV}$$

Transition from $n=4 \rightarrow n=2$

C) Say the atom is not hydrogen, and the transition is from $n=4$ to $n=3$
What atom would this be?

$$\Rightarrow E_n = -13.6 \frac{Z^2}{n^2}$$

$$E_4 = -0.85 Z^2 \quad \frac{1}{n_f^2} - \frac{1}{n_i^2}$$

$$E_3 = -1.51 Z^2$$

$$E_4 - E_3 = 0.66 Z^2$$

$$2.55 = 0.66 Z^2$$

$$Z \sim 2$$

likely He^+

D) Assume it emits like a blackbody. Find temperature.

$$\Rightarrow \lambda_{\text{peak}} = \frac{b}{T}$$

$$T = \frac{3 \times 10^{-3} \text{ mK}}{4861.1 \text{ \AA}} = 6171 \text{ K}$$

Typically quasars or HII regions have temperature $\sim 10^4 \text{ K}$.